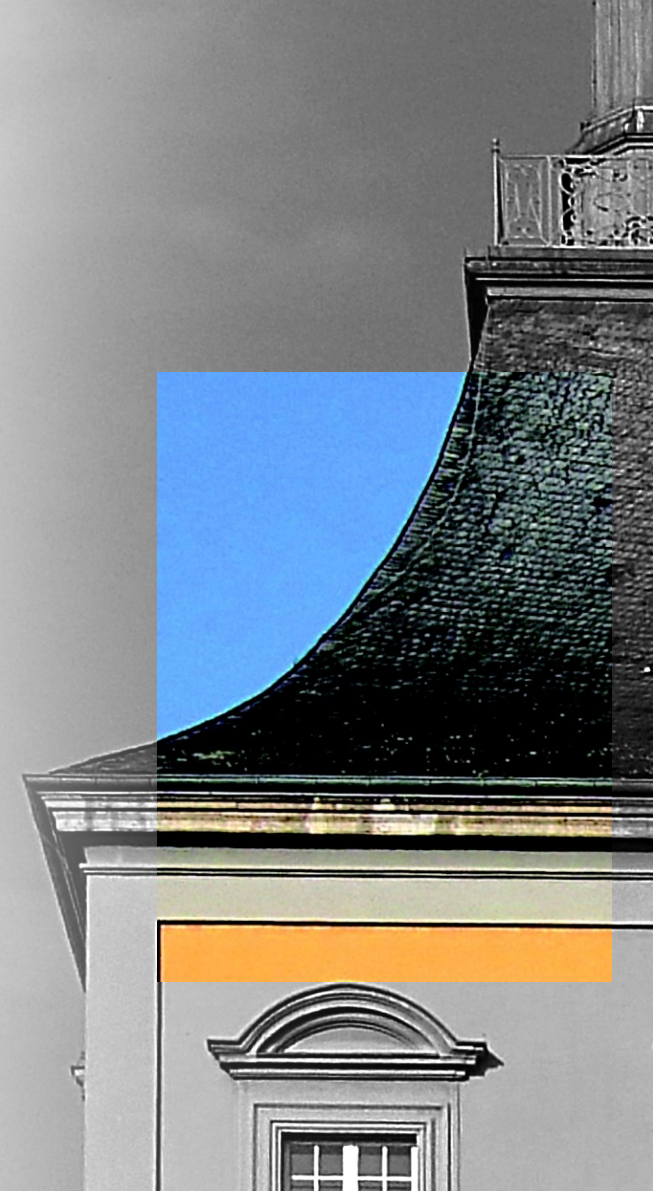


Electrochemical copper deposition process

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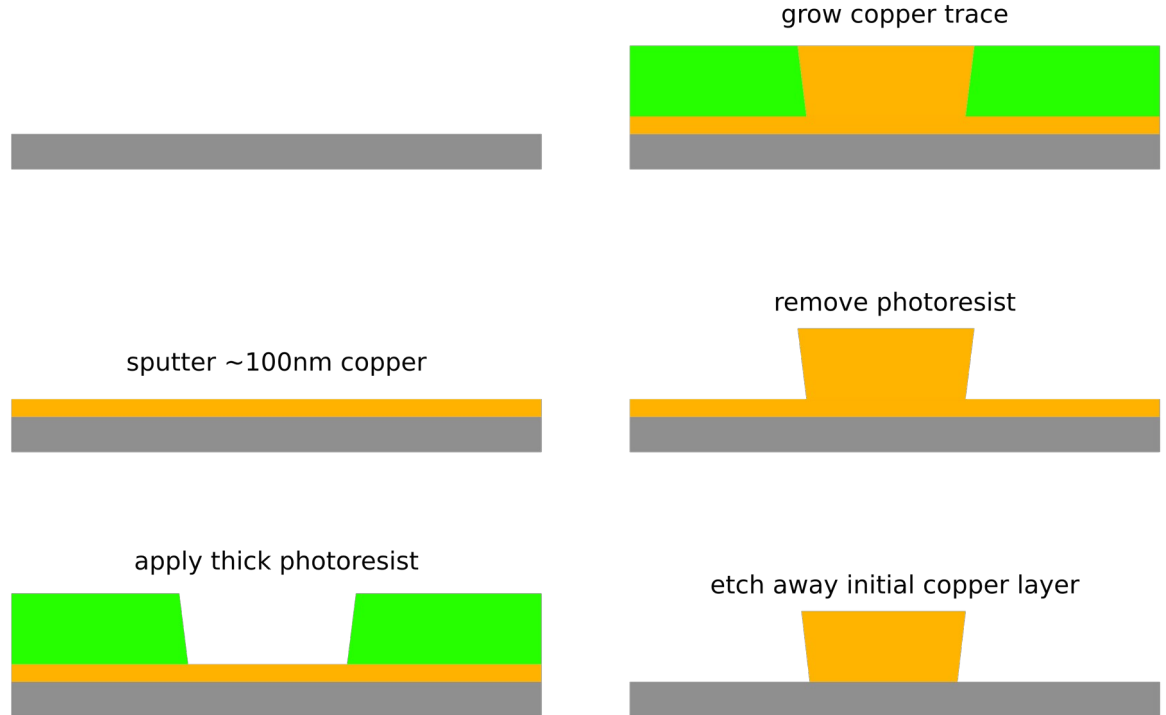
Motivation

We should get a electrochemical copper deposition process:

- Useful for thick layers of metal
- Copper has better conductivity than Al
- Less stress for metalization process
- Used for All-Silicon Modules as IZM does not have thick Al process
 - Useful to have comparable process to create test structures to characterize technology
 - Much, much cheaper to build structures by ourselves
- Useful for future projects that absolutely need copper
- First step for Through Silicon Vias (TSV)

Copper trace process:

- Deposit thin layer (seed layer of ~100nm) of copper on substrate
- Apply thick photoresist (at least in thickness of trace) on copper layer
- Grow copper trace electrochemically in structured photoresist
- Remove photoresist
- Etch away seed layer thickness



Electrochemical deposition (ECD)

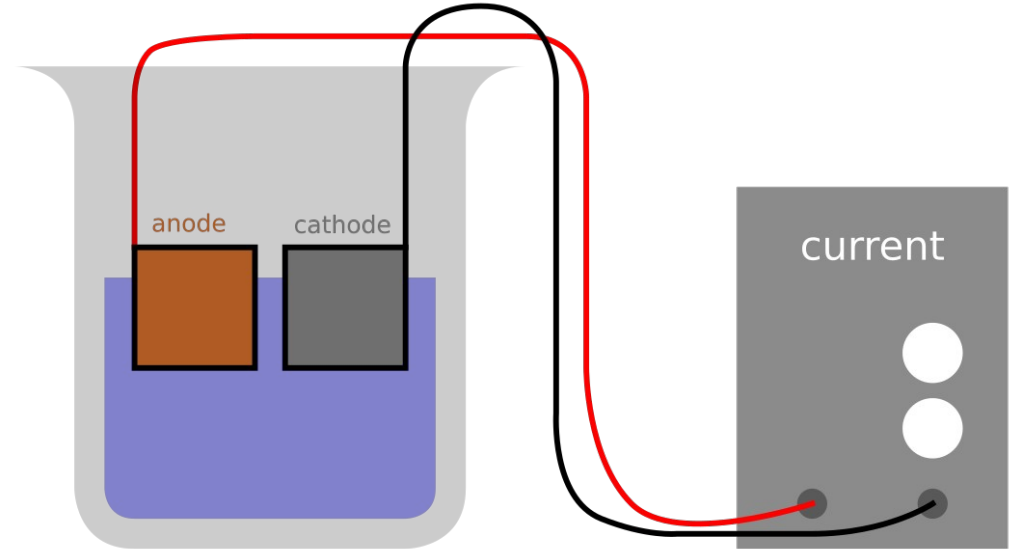
- Non-toxic **copper sulfate** bath for deposition
- Copper sulfate + sulfuric acid forming ionized species in solution
 - Solution is highly conductive
 - Sulfuric acid can attack sputtered metals
 $(2\text{Al} + 3(\text{H}_2\text{SO}_4) \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{H}_2)$
 - Investigation of effect necessary! Possibly other seed metals necessary (chromium, copper)
- Current density of 215 – 1075 A/m²
 - Lower currents can cause copper sulfate build up
 - High currents can destroy copper anode → rough deposit
- Electrode efficiencies should be close to 100% (need investigation if Al cathode is usable)
- Electrolysis reaction:
 - Cathode reaction: $\text{Cu}^{2+}(\text{aq}) + 2\text{e}^- \rightarrow \text{Cu}(\text{s})$
 - Anode reaction: $2\text{H}_2\text{O}(\text{l}) \rightarrow \text{O}_2(\text{g}) + 4\text{H}^+(\text{aq}) + 4\text{e}^-$

What needs to be investigated

- Can we grow copper on aluminum (or do we need a different target like copper or chromium)
- Adhesion on wafer/aluminum/photoresist
- Copper growth rate
- Copper growth uniformity
- Investigate dependence of
 - Pretreatment of aluminum electrode
 - Current
 - Concentration of solution
 - Initial layer thickness
 - Pulse current deposition
 - Other possible parameters

Where to start

- Sputter 2cm x 2cm chips with 100nm Al (also good test for new Al targets)
- Build small ECD setup in ISO5 wetbench
 - Lab PSU
 - Glasware
 - Copper Anode
 - Cables
 - Cathode holder
 - Stirrer
 - Copper sulfate solution
- Investigate deposition properties for pure Copper sulfate solution first (expecting ~10mA)
- Add Sulfuric Acid to solution (mix new solution) and investigate metalization process (expecting ~100mA)



Outlook

- Once process was initially tested, try improvements:
 - Development of procedures
 - Variation of additives (Chloride, Thiourea, wetting agent, Dextrin, Molasses)
 - Improve setup mechanically
- Try premixed solutions for commercial use

- Once process is established:
 - Investigate TSV capabilities (Maybe with BASF solution or self mixed solution)